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## Key indicators

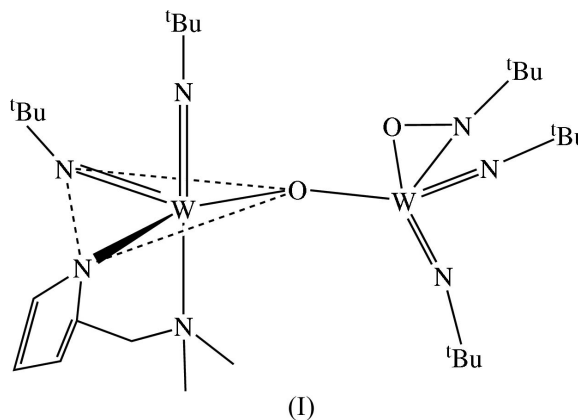
Single-crystal X-ray study  
 $T = 150$  K  
Mean  $\sigma(C-C) = 0.007$  Å  
 $R$  factor = 0.026  
 $wR$  factor = 0.063  
Data-to-parameter ratio = 25.4For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.**Bis(*tert*-butylimido)[2-(dimethylaminomethyl)-  
pyrrol-1-yl]tungsten(VI)- $\mu$ -oxo-bis(*tert*-butylimido)-  
( $\eta^2$ -*N,O*-*tert*-butylnitroso)tungsten(VI)**The structure of the title compound,  $[W_2(C_4H_9N)_4(C_7H_{11}N_2)-(C_4H_{10}NO)O]$ , was determined at 150 K. It is an asymmetrical binuclear compound with a  $\mu$ -oxo bridge linking two tungsten(VI) centers.

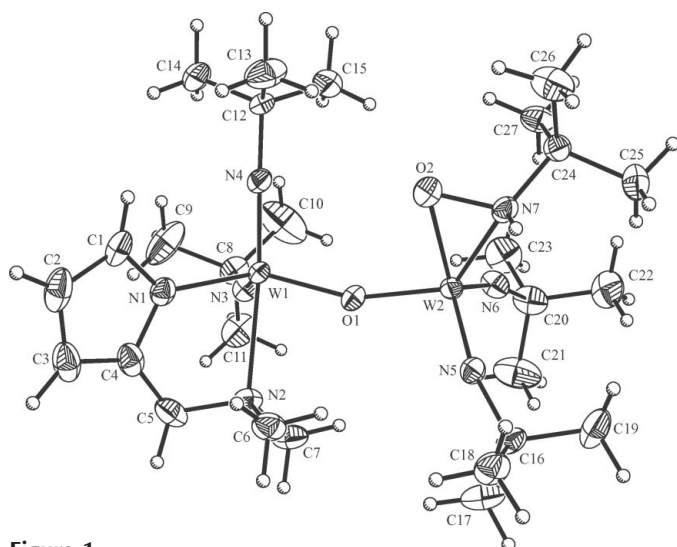
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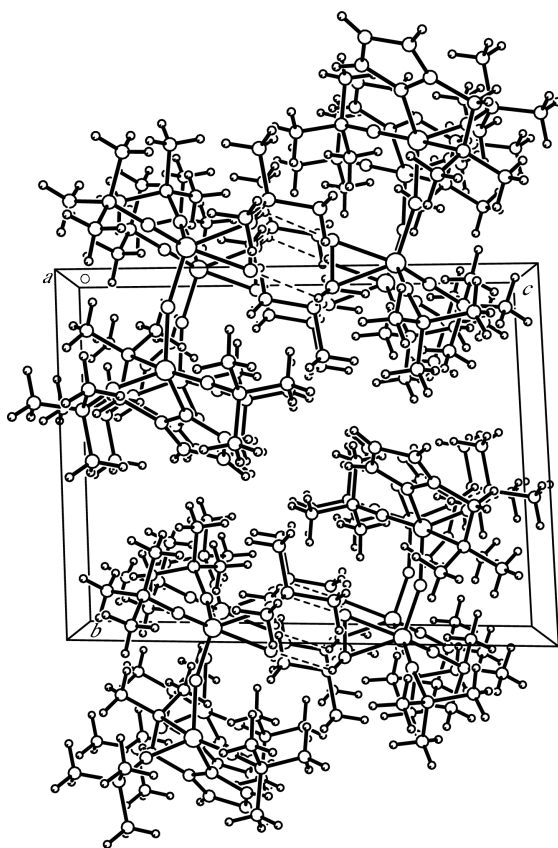
Online 18 March 2005

## Comment

Organoimide chemistry has been investigated extensively and reviewed periodically (Nugent & Haymore, 1980; Nugent & Mayer, 1988). The organoimide compounds involving monoanionic monodentate ligands have been commonly observed. However, monoanionic bidentate organoimide complexes containing substituted pyrrole ligands are less reported (Huang *et al.*, 2000). Currently, we are investigating the chemistry of tungsten–imide complexes and bidentate 2-(dimethylaminomethyl)pyrrole (LH) and we present here the structure of the title compound, (I).The title compound is a binuclear tungsten(VI) complex in which one of the tungsten centers involves a bidentate 2-(dimethylaminomethyl)pyrrolyl and two *tert*-butylimido ligands. The other tungsten center involves two *tert*-butylimido ligands and, intriguingly, an  $\eta^2$ -*N,O*-*tert*-butylnitroso ligand. The two tungsten centers are linked by a  $\mu$ -oxo bridge with a bond angle of 159.69 (16) Å (Fig. 1).In Fig. 1, the W atom on the left (W1) is in a distorted trigonal bipyramidal geometry, whereas that on the right (W2) is in a distorted tetrahedral geometry with the  $\eta^2$ -*N,O*-*tert*-butylnitroso ligand occupying one coordination site. The asymmetrical binuclear compound has no molecular symmetry.Centrosymmetric pairs of intermolecular hydrogen bonds of the type N–H $\cdots$ O are present in the crystal structure (Table 1). Fig. 2 shows the crystal packing.



**Figure 1**  
The structure of (I), showing 50% probability displacement ellipsoids for the non-H atoms.



**Figure 2**  
The packing of (I), viewed along the *a* axis. Dashed lines indicate hydrogen bonds.

## Experimental

Crystals of the title complex were obtained from the reaction of 2-(dimethylaminomethyl)pyrrole (0.66 g, 5 mmol) and bis(*tert*-butylimido)bis(*tert*-butylamido)tungsten (2.5 g, 5 mmol) in diethyl ether at 195 K for 12 h followed by repeated recrystallization from a diethyl ether solution.

## Crystal data

$[\text{W}_2(\text{C}_4\text{H}_9\text{N})_4(\text{C}_7\text{H}_{11}\text{N}_2)(\text{C}_4\text{H}_{10}\text{NO})\text{O}]$   
 $M_r = 879.50$   
 Triclinic,  $P\bar{1}$   
 $a = 9.6263(3) \text{ \AA}$   
 $b = 12.0153(3) \text{ \AA}$   
 $c = 15.6375(4) \text{ \AA}$   
 $\alpha = 86.334(2)^\circ$   
 $\beta = 79.479(2)^\circ$   
 $\gamma = 83.175(2)^\circ$   
 $V = 1764.02(8) \text{ \AA}^3$

$Z = 2$   
 $D_x = 1.656 \text{ Mg m}^{-3}$   
 Mo  $K\alpha$  radiation  
 Cell parameters from 5880 reflections  
 $\theta = 2.3\text{--}28.8^\circ$   
 $\mu = 6.55 \text{ mm}^{-1}$   
 $T = 150(2) \text{ K}$   
 Block, colorless  
 $0.53 \times 0.34 \times 0.27 \text{ mm}$

## Data collection

Bruker SMART APEX-II diffractometer  
 $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Sheldrick, 2003)  
 $T_{\min} = 0.080$ ,  $T_{\max} = 0.171$   
 25718 measured reflections

9142 independent reflections  
 7668 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.029$   
 $\theta_{\max} = 28.9^\circ$   
 $h = -12 \rightarrow 13$   
 $k = -16 \rightarrow 16$   
 $l = -21 \rightarrow 21$

## Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.026$   
 $wR(F^2) = 0.063$   
 $S = 1.04$   
 9142 reflections  
 360 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0236P)^2 + 3.8003P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.002$   
 $\Delta\rho_{\max} = 3.04 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -1.90 \text{ e \AA}^{-3}$

**Table 1**

Hydrogen-bonding geometry ( $\text{\AA}$ ,  $^\circ$ ).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N7—H7...O2 <sup>i</sup>	0.93	2.53	3.220 (4)	132

Symmetry code: (i)  $1 - x, -y, 1 - z$ .

All H atoms were positioned geometrically ( $\text{C—H} = 0.95\text{--}0.99 \text{ \AA}$ ) and refined with a riding model, with  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$  for methyl H atoms and  $1.2U_{\text{eq}}(\text{C})$  for all other H atoms. The maximum and minimum electron-density peaks are 0.74 and 0.62  $\text{\AA}^{-3}$ , respectively, from atom W2.

Data collection: APEX2 (Bruker, 2004); cell refinement: APEX2; data reduction: APEX2; program(s) used to solve structure: SHELXTL (Sheldrick, 1998); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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## References

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